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First total synthesis of herbarumin \mathbf{III}^{\bigstar}

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Abstract—An asymmetric 10-step total synthesis of herbarumin III in 24% overall yield is described using ring-closing metathesis as the key step.

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The interesting biological activities of herbarumin I (2) and II (3) and the reinvestigation of the fermentation broth and mycelium of the fungus *Phoma herbarum* led to the isolation of a new phytotoxic nonenolide (7R,9R)-7-hydroxy-9-propyl-5-nonen-9-olide (herbarumin III, 1), along with herbarumin I (2) and II (3).¹ The structure of 1 was elucidated by spectroscopic methods combined with molecular modeling. Compounds 1 to 3 interacted with bovine-brain calmodulin and inhibited the activation of the calmodulin-dependent enzyme camp phosphodiesterase.



In general, synthetic approaches to lactones have focused mainly on the use of fragmentation/ring expansion reactions and on lactonization strategies in order to build the lactone ring.² Ring-closing metathesis (RCM)³ is a modern synthetic protocol to access various lactone rings. Total syntheses of herbarumin I (2)⁴ and herbarumin II (3)⁵ have been reported recently following RCM or Nozaki–Hiyama–Kishi reactions as the key

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steps. Our synthetic strategy for the total synthesis of herbarumin III (1) also hinges on ring-closing metathesis (RCM) for lactone ring formation with the required chiral aliphatic diol envisaged to arise from D-glucose.

We began our synthesis with D-glucose as the starting material which was converted to its diacetonide and corresponding xanthate derivative followed by deoxygenation under the Barton-McCombie protocol using *n*-Bu₃SnH and a catalytic amount of AIBN in toluene under reflux conditions to provide the 3-deoxyglucose derivative 4 in 58% yield over three steps.^{6a,b} Regioselective monohydrolysis of the 5,6-O-isopropylidene moiety of 4 with 0.8% H₂SO₄ in MeOH at ambient temperature afforded the diol 5 in 84% yield. Introduction of the *n*-propyl group at the C-4 position was achieved in three steps via oxidative cleavage of diol 5 using silica gel supported⁷ NaIO₄ and a two carbon homologation using ethyltriphenylphosphonium bromide and *n*-BuLi in anhydrous THF at -78 °C, followed by hydrogenation using 10% Pd/C to afford the *n*-propyl derivative 8 in 72% overall yield. Deprotection of the 1,2-O-isopropylidene group of compound 8 with 20% acetic acid in H₂O and a catalytic amount of H₂SO₄ afforded the lactol 9 in 93% yield, which after work up, was subjected to Wittig methylenation⁸ with in situ generated methylenetriphenylphosphorane to afford the olefinic intermediate 10. Selective protection of the allylic hydroxyl group as its PMB-ether and esterification of the other hydroxyl group with 5-hexenoic acid⁹ under Yamaguchi et al. conditions¹⁰ at room temperature afforded the desired ring-closing metathesis precursor 12 in 82% yield.

The ring-closing metathesis was then investigated using Grubbs' first and second generation catalysts. In our

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Scheme 1. Reagents, conditions and yields: (a) 0.8% H₂SO₄, MeOH, rt, 12 h, (84%); (b) silica gel supported NaIO₄, CH₂Cl₂, rt, 30 min (95%); (c) Br⁻P⁺Ph₃CH₂CH₃, *n*-BuLi, THF, -78 to 0 °C, 3 h (82%); (d) H₂, Pd/C, 1 bar, rt, 3 h (92%); (e) 20% AcOH in H₂O, concd H₂SO₄ (catalytic), reflux, 6 h (93%); (f) I⁻P⁺Ph₃CH₃, *n*-BuLi, THF, 0 °C to rt, 12 h (76%); (g) PMBCl, NaH, DMF, 0 °C, 1 h, (94%); (h) 5-hexenoic acid, 2,4,6-trichlorobenzoyl chloride, DMAP, THF, 11 in THF, 0 °C to rt, 4 h (82%); (i) DDQ, CH₂Cl₂/H₂O (18:1), rt, 30 min (94%); (j) (i) RuCl₂(=CHPh)(PCy₃)₂, CH₂Cl₂, reflux, 4 days, (36%) (52% starting material recovered), (ii) RuCl₂(=CHPh)(PCy₃)(IEMS), CH₂Cl₂, reflux, 16 h, (78%).

hands, RCM of compound **12** led to the formation of the homodimer as the major product. With the use of precursor **13** for RCM, we could almost completely suppress the homodimerization. The reaction of **13** with the first generation Grubbs' catalyst in CH_2Cl_2 (high dilution) under reflux conditions gave the natural product **1** in 36% yield as the only product after 4 days with 52% recovery of the starting material, while the second generation catalyst provided the final compound **1** in 78% yield (Scheme 1). Spectral and analytical data obtained for **1** were consistent with those reported in the literature.¹

In conclusion, we have completed the first total synthesis of the biologically active natural product herbarumin III (1) starting from D-glucose.

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